Experiment of Inorganic & Analytical Chemistry

Couse code: 215204

Couse status: Basic course

Applied: Food Science and Engineering(class hours: 51 in total)、 Chemical Engineering and Technology(class hours: 51 in total)、 Biology Engineering(class hours: 51 in total)、 Pharmacy Engineering(class hours: 51 in total)、 Materials Science and Engineering(class hours: 51 in total)、 Chemical Engineering and Technology(bachelor from certificate) (class hours: 48 in total) Class Hours: 51 in total Credit Value: 1.5 in total Writer: Li hui

1. Function

Experiment of Inorganic & Analytical Chemistry covers principles and applications of chemical laboratory techniques, including preparation and analysis of chemical materials, measurement of pH, gas and liquid chromatography, visible-ultraviolet spectrophotometry, infrared spectroscopy, kinetics, data analysis, and elementary synthesis. It adjusts for the basic education of such specialty: Food Science and Engineering、 Chemical Engineering and Technology、 Materials Science and Engineering、 Biology Engineering、 Pharmacy Engineering.

2. Objectives:

The program is to prepare students to master theory and technology of Inorganic & Analytical Chemistry, especial in the skills of Experiment of Inorganic & Analytical Chemistry, and equip them with basic knowledge and applications skills of Inorganic & Analytical Chemistry. The program prepares students for future careers in the fields of engineering design, science research, technical development, education, and management, as well as in other related fields, especially in Experiment of Inorganic & Analytical Chemistry of the 21st century.

number	item	Class	regimentation	requirement	students
		Hours			per
					team
1	Safety Practices in the	2	demo	compulsory	1
	Chemistry Laboratory				
2	The Purification of Rough	3	verification	compulsory	1
	Table salt				
3	The Preparation of Mohr salt	5	integration	compulsory	1
	From Waster Iron Particles				
4	Preparation of Buffer Solution	3	verification	compulsory	2
	and Its Properties				
5	The Operation Exercise of	3	verification	compulsory	1
	Volume Analysis				

3. Contents in total:

6	The Measurement of	3	verification	compulsory	1
	Concentrations of Hcl(aq) and				
	NaOH(aq) by Neutral Titration				
7	The Measurement of the	3	verification	compulsory	1
	Concentrations of the NaOH				
	and Na ₂ CO ₃ in the mixed				
	alkaline liquid				
8	The preparation and	3	verification	compulsory	1
	standardization of the standard				
	solution of EDTA				
9	The Measurement of the	3	verification	compulsory	1
	Degree of hardness of water				
10	The preparation and	3	verification	compulsory	1
	standardization of the standard				
	solution of $Na_2S_2O_3$				
11	The Measurement of the	3	verification	compulsory	1
	Copper concentration of CuSO ₄				
12	The Measurement of Iron in	5	verification	compulsory	3
	Iron(II) Phenanthroline ion by				
	Spectrophotometric Method				
13	The preparation and	3	verification	compulsory	1
	standardization of the standard				
	solution of Silver nitrate				
14	The preparation and	9	integration	compulsory	2
	Component Analysis of				
	$[Co(NH_3)_6]Cl_3$				

4. Assessment:

Assessments(70%): including preparation, practical and papers of experiments; Final test(30%).

5. Essential Reading:

Li, juqing, Yu, yuanzhi, zhang, liqing, zhang, peizhi. Experiment of Inorganic & Analytical Chemistry, zhe jiang University Of Science and Technology Press, Jul. 2003;

6. References:

1) Chen,huapu、 Experiment of Inorganic & Analytical Chemistry、 Chemistry Technology Press,1998;

2) Ni, jing'an, Inorganic & Analytical Chemistry, Chemistry Technology Press, 1999;

3) Cui,xuegui、zhang,xiaoli. Experiments of Basic chemistry- Experiment of Inorganic & Analytical Chemistry、Chemistry Technology Press,Jul.2003;